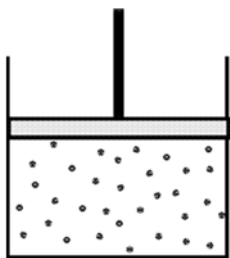


Volume and Pressure
Temperature is constant

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$



The relationship is (circle one)

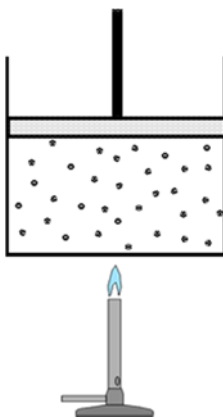
Inverse / Direct



Volume and Temperature
Pressure is constant

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Temperature must be in _____
K = °C + 273



The relationship is (circle one)

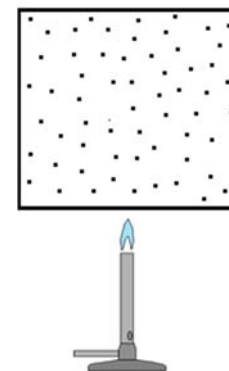
Inverse / Direct



Temperature and Pressure
Volume is constant

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Temperature must be in _____
K = °C + 273



The relationship is (circle one)

Inverse / Direct

